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## 2,2'-Spirobi(1,3-benzodithiole)

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The X-ray structure analysis of $2,2^{\prime}$-spirobi(1,3-benzodithiole), $\mathrm{C}_{13} \mathrm{H}_{8} \mathrm{~S}_{4}$, has been performed. The molecule has crystallographic twofold rotation symmetry, the axis passing through the spiro-C atom. The four S atoms are arranged around the spiro-C atom in two almost orthogonal $\mathrm{CS}_{2}$ planes. However, because of large bending of the two five-membered rings, close contact is present between two connected C atoms on the benzo group of each 1,3-benzodithiole ring and one $S$ atom on the other 1,3-benzodithiole ring.

## Comment

In the title compound, (I), the $\mathrm{S} 1-\mathrm{C} 1, \mathrm{~S} 1-\mathrm{C} 2, \mathrm{~S} 2-\mathrm{C} 1, \mathrm{~S} 2-$ $\mathrm{C} 7, \mathrm{~S} 1^{\mathrm{i}}-\mathrm{C} 1, \mathrm{~S} 1^{\mathrm{i}}-\mathrm{C} 2^{\mathrm{i}}, \mathrm{S} 2^{\mathrm{i}}-\mathrm{C} 1$ and $\mathrm{S} 2^{\mathrm{i}}-\mathrm{C} 7^{\mathrm{i}}$ bond distances [symmetry code: (i) $y, x, 1-z$ ] are in the range $1.77-1.84 \AA$, which are near to the values reported so far (1.80-1.81 $\AA$; Sutton, 1958, 1965). The C1-S1-C2, C1-S2-C7, C1-S1 $1^{i}-$ $\mathrm{C} 2^{\mathrm{i}}$ and $\mathrm{C} 1-\mathrm{S} 2^{\mathrm{i}}-\mathrm{C} 7^{\mathrm{i}}$ bond angles are in the range $94-96^{\circ}$ and slightly smaller than the normal values ( $97-99^{\circ}$; Sutton, 1958, 1965). The $\mathrm{S} 1-\mathrm{C} 1-\mathrm{S} 2$ and $\mathrm{S} 1^{\mathrm{i}}-\mathrm{C} 1-\mathrm{S} 2^{\mathrm{i}}$ planes make a dihedral angle of $93.8(1)^{\circ}$. The five-membered $\mathrm{C} 1-\mathrm{S} 1-\mathrm{C} 2-$ $\mathrm{C} 7-\mathrm{C} 2$ and $\mathrm{C} 1-\mathrm{S} 1^{\mathrm{i}}-\mathrm{C} 2^{\mathrm{i}}-\mathrm{C} 7^{\mathrm{i}}-\mathrm{C} 2^{\mathrm{i}}$ rings have an envelopelike conformation, as seen from the dihedral angles between

(I)
the $\mathrm{C} 1 / \mathrm{S} 1 / \mathrm{S} 2$ and $\mathrm{S} 1 / \mathrm{S} 2 / \mathrm{C} 7 / \mathrm{C} 2$ planes [35.0 (1) ${ }^{\circ}$ ], resulting in a dihedral angle of $79.2(1)^{\circ}$ between the two benzene rings. As the result of large bending of the two five-membered rings, C2 and $\mathrm{C} 2^{\mathrm{i}}$ or C 7 and $\mathrm{C} 7^{\mathrm{i}}$ atom on the benzo group of one $1,3-$ benzodithiole ring and the S 2 or $\mathrm{S} 2^{\mathrm{i}}$ atom on the other 1,3benzodithiole ring become closer to each other. In particular, the closest contact can be seen between S 2 and $\mathrm{C}^{\mathrm{i}}$ atoms
[3.58 (1) $\AA$ ]. The contact distance is near to the sum of van der Waals' radii of $\mathrm{C}(1.75 \AA$ ) and S atoms ( $1.85 \AA$; Pauling, 1960).

The ${ }^{1} \mathrm{H}$ NMR spectrum of (I) in DMF- $d_{7}$ showed each of two kinds of signals due to the ortho-protons $\left(\mathrm{H} 3 / \mathrm{H}^{\mathrm{i}}\right.$ and $\mathrm{H} 6 /-$ $\left.\mathrm{H} 6^{\mathrm{i}}\right)$ at $\delta 7.26(2 \mathrm{H}, d d, J=2.57,3.33 \mathrm{~Hz})$ and $7.46(2 \mathrm{H}, d d, J=$ $2.57,3.33 \mathrm{~Hz}$ ), and to the meta-protons $\left(\mathrm{H} 4 / \mathrm{H} 4^{\mathrm{i}}\right.$ and $\left.\mathrm{H} 5 / \mathrm{H} 5^{\mathrm{i}}\right)$ at $\delta 7.27(2 \mathrm{H}, t, J=3.33 \mathrm{~Hz})$ and $7.44(2 \mathrm{H}, t, J=3.33 \mathrm{~Hz})$ on the benzo groups. The different chemical shifts between $\mathrm{H} 3 / \mathrm{H} 3^{i}$ and $\mathrm{H} 6 / \mathrm{H} 6^{i}$ protons and between $\mathrm{H} 4 / \mathrm{H} 4^{\mathrm{i}}$ and $\mathrm{H} 5 / \mathrm{H} 5^{\mathrm{i}}$ protons are caused by the magnetic anisotropy due to the sulfur lone pairs in close proximity to $\mathrm{H} 3 / \mathrm{H} 3^{\mathrm{i}}$ and $\mathrm{H} 4 / \mathrm{H} 4^{\mathrm{i}}$ protons, as seen from the crystal structure. On the other hand, only three kinds of signals ( $138.22,129.01$ and 124.64 p.p.m.) due to the benzo C atoms were observed together with the spiro-carbon signal (90.01 p.p.m.) in the ${ }^{13} \mathrm{C}$ NMR spectrum. This result clearly suggests that also in solution the five-membered rings maintain an envelope-like conformation without rapid flipping on the NMR time-scale (Horn et al., 1993; Setaka et al., 1999).

## Experimental

The title compound was first obtained as a by-product ( $<5 \%$ yield) in the thermal decomposition of 2-( $n$-amyloxybenzo)-1,3-dithiole (Nakayama, 1975). In order to improve the yield, the following method was now used. 2-Methylthio-1,3-benzodithiolium tetrafluoroborate was reacted with an equimolar amount of benzene-1,2dithiol in the presence of a catalytic amount of $p$-toluenesulfonic acid in 1,2-dichloroethane at 356 K for 3 h . The crude product was purified by silica-gel column chromatography with an eluent of $n$-hexane-
/benzene ( $v / v=3: 1$ ), followed by recrystallization from benzene $/ n$ hexane to give colourless crystals of (I) (m.p. 393 K ) in $23 \%$ yield.

Crystal data
$\mathrm{C}_{13} \mathrm{H}_{8} \mathrm{~S}_{4}$
$M_{r}=292.45$
Tetragonal, $P 4_{1} 2_{1} 2$
$a=6.287$ (6) $\AA$
$c=31.93(1) \AA$
$V=1261.9(8) \AA^{3}$
$Z=4$
$D_{x}=1.539 \mathrm{Mg} \mathrm{m}^{-3}$

## Mo $K \alpha$ radiation

Cell parameters from 25 reflections
$\theta=10.4-11.9^{\circ}$
$\mu=0.723 \mathrm{~mm}^{-1}$
$T=293.2 \mathrm{~K}$
Prismatic, colorless
$0.20 \times 0.20 \times 0.10 \mathrm{~mm}$

## Data collection

Rigaku AFC-7R diffractometer $\omega-2 \theta$ scans
934 measured reflections
934 independent reflections
755 reflections with $F^{2}>2 \sigma\left(F^{2}\right)$
$\theta_{\text {max }}=27.33^{\circ}$

## Refinement

Refinement on $F$
$R=0.0347$
$w R=0.0453$
$S=1.325$
803 reflections $[I>0.7 \sigma(I)]$
77 parameters

$$
\begin{aligned}
& h=0 \rightarrow 8 \\
& k=0 \rightarrow 5 \\
& l=0 \rightarrow 41 \\
& 3 \text { standard reflections } \\
& \quad \text { every } 150 \text { reflections } \\
& \quad \text { intensity decay: } 0.70 \%
\end{aligned}
$$

H -atom parameters not refined
$w=1 /\left[\sigma^{2}\left(F_{o}\right)+0.00040\left|F_{o}\right|^{2}\right]$
$(\Delta / \sigma)_{\max }=0.039$
$\Delta \rho_{\text {max }}=0.17 \mathrm{e}^{\AA^{-3}}$
$\Delta \rho_{\min }=-0.17 \mathrm{e}^{-3}$

Table 1
Selected geometric parameters $\left(\AA^{\circ},{ }^{\circ}\right)$.

|  |  |  |  |
| :--- | :---: | :--- | ---: |
| S1-C1 | $1.821(3)$ | $\mathrm{C} 2-\mathrm{C} 7$ | $1.407(5)$ |
| $\mathrm{S} 1-\mathrm{C} 2$ | $1.77(1)$ | $\mathrm{C} 3-\mathrm{C} 4$ | $1.404(9)$ |
| $\mathrm{S} 2-\mathrm{C} 1$ | $1.843(3)$ | $\mathrm{C} 4-\mathrm{C} 5$ | $1.388(6)$ |
| $\mathrm{S} 2-\mathrm{C} 7$ | $1.768(10)$ | $\mathrm{C} 5-\mathrm{C} 6$ | $1.380(8)$ |
| $\mathrm{C} 2-\mathrm{C} 3$ | $1.391(8)$ | $\mathrm{C} 6-\mathrm{C} 7$ | $1.391(8)$ |
|  |  |  |  |
| $\mathrm{C} 1-\mathrm{S} 1-\mathrm{C} 2$ | $94.7(2)$ | $\mathrm{S} 1-\mathrm{C} 2-\mathrm{C} 7$ | $116.5(3)$ |
| $\mathrm{C} 1-\mathrm{S} 2-\mathrm{C} 7$ | $95.2(2)$ | $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 7$ | $120.1(4)$ |
| $\mathrm{S} 1-\mathrm{C} 1-\mathrm{S} 1^{\mathrm{i}}$ | $112.5(3)$ | $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | $118.7(5)$ |
| $\mathrm{S} 1-\mathrm{C} 1-\mathrm{S} 2$ | $107.24(8)$ | $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5$ | $120.5(4)$ |
| $\mathrm{S} 1-\mathrm{C} 1-\mathrm{S} 2^{\mathrm{i}}$ | $108.0(2)$ | $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ | $121.2(4)$ |
| $\mathrm{S} 1-\mathrm{C} 1^{\mathrm{i}}-\mathrm{S} 2$ | $107.2(4)$ | $\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 7$ | $119.0(5)$ |
| $\mathrm{S} 1-\mathrm{C} 1^{\mathrm{i}}-\mathrm{S} 2^{\mathrm{i}}$ | $108.0(1)$ | $\mathrm{S} 2-\mathrm{C} 7-\mathrm{C} 2$ | $115.2(3)$ |
| $\mathrm{S} 2-\mathrm{C} 1-\mathrm{S} 2^{\mathrm{i}}$ | $114.0(4)$ | $\mathrm{S} 2-\mathrm{C} 7-\mathrm{C} 6$ | $124.2(4)$ |
| $\mathrm{S} 1-\mathrm{C} 2-\mathrm{C} 3$ | $123.3(4)$ | $\mathrm{C} 2-\mathrm{C} 7-\mathrm{C} 6$ | $120.6(4)$ |
|  |  |  |  |
| $\mathrm{S} 1-\mathrm{C} 1-\mathrm{S} 1^{\mathrm{i}}-\mathrm{C} 2^{\mathrm{i}}$ | $150.1(1)$ | $\mathrm{C} 1-\mathrm{S} 1^{\mathrm{i}}-\mathrm{C} 2^{\mathrm{i}}-\mathrm{C} 3^{\mathrm{i}}$ | $159.9(4)$ |
| $\mathrm{S} 1-\mathrm{C} 1-\mathrm{S} 2-\mathrm{C} 7$ | $-31.0(2)$ | $\mathrm{C} 1-\mathrm{S} 1^{\mathrm{i}}-\mathrm{C} 2^{\mathrm{i}}-\mathrm{C} 7^{\mathrm{i}}$ | $-21.3(3)$ |
| $\mathrm{S} 1-\mathrm{C} 1-\mathrm{S} 2^{\mathrm{i}}-\mathrm{C} 7^{\mathrm{i}}$ | $-152.5(2)$ | $\mathrm{C} 1-\mathrm{S} 2-\mathrm{C} 7-\mathrm{C} 2$ | $18.7(3)$ |
| $\mathrm{S} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | $176.7(3)$ | $\mathrm{C} 1-\mathrm{S} 2-\mathrm{C} 7-\mathrm{C} 6$ | $-162.9(3)$ |
| $\mathrm{S} 1-\mathrm{C} 2-\mathrm{C} 7-\mathrm{S} 2$ | $1.7(4)$ | $\mathrm{C} 1-\mathrm{S} 2^{\mathrm{i}}-\mathrm{C} 7^{\mathrm{i}}-\mathrm{C} 2^{\mathrm{i}}$ | $18.7(3)$ |
| $\mathrm{S} 1-\mathrm{C} 2-\mathrm{C} 7-\mathrm{C} 6$ | $-176.8(3)$ | $\mathrm{C} 1-\mathrm{S} 2^{\mathrm{i}}-\mathrm{C} 7^{\mathrm{i}}-\mathrm{C} 6^{\mathrm{i}}$ | $-162.9(3)$ |
| $\mathrm{S} 2-\mathrm{C} 1-\mathrm{S} 1-\mathrm{C} 2$ | $31.5(2)$ | $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5$ | $0.1(6)$ |
| $\mathrm{S} 2-\mathrm{C} 1-\mathrm{S} 1^{\mathrm{i}}-\mathrm{C} 2^{\mathrm{i}}$ | $-91.8(2)$ | $\mathrm{C} 2-\mathrm{C} 7-\mathrm{C} 6-\mathrm{C} 5$ | $0.0(6)$ |
| $\mathrm{S} 2-\mathrm{C} 1-\mathrm{S} 2^{\mathrm{i}}-\mathrm{C} 7^{\mathrm{i}}$ | $88.4(4)$ | $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 7-\mathrm{C} 6$ | $2.0(5)$ |
| $\mathrm{S} 2-\mathrm{C} 7-\mathrm{C} 2-\mathrm{C} 3$ | $-179.5(3)$ | $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ | $2.0(6)$ |
| $\mathrm{S} 2-\mathrm{C} 7-\mathrm{C} 6-\mathrm{C} 5$ | $-178.3(3)$ | $\mathrm{C} 4-\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 7$ | $-2.0(6)$ |
| $\mathrm{C} 1-\mathrm{S} 1-\mathrm{C} 2-\mathrm{C} 3$ | $159.9(3)$ | $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 7$ | $-2.0(6)$ |
| $\mathrm{C} 1-\mathrm{S} 1-\mathrm{C} 2-\mathrm{C} 7$ | $-21.3(3)$ |  |  |

Symmetry code: (i) $y, x, 1-z$.

Data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1985); cell refinement: MSC/AFC Diffractometer Control Software; data reduction: TEXSAN (Molecular Structure Corporation, 1999); program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: TEXSAN; software used to prepare material for publication: TEXSAN.

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